Name:

Provide a molecular orbital diagram for Be₂⁺.
 Predict the bond order and the magnetic class of this diatomic ie paramagnetic or diamagnetic.

2. Using **Band Theory** describe how a p-type semiconductor differs from an n-type semiconductor.

3. Provide **IUPAC names** for the following molecules:

$$\begin{array}{c|c} \mathsf{CH_3} & \mathsf{O} & \mathsf{CH_3} \\ \mathsf{CH_3}\mathsf{CHCH_2}\mathsf{C} & \mathsf{CH_3}\mathsf{CCH_2}\mathsf{CH_2}\mathsf{CHCH_3} \\ \mathsf{H} & \mathsf{O} & \mathsf{O} \\ \mathsf{CH_3}\mathsf{CCH_2}\mathsf{CH_2}\mathsf{CHCH_3} \end{array}$$

4. Draw **skeletal structures** for the following molecules:

4-iodo-4-ethyl-2-methylhexane

benzyl bromide

5. Provide **IUPAC names** for the following molecules:

$$CH_3$$
 H
 $(CH_2)_4CH_2OH$
 Br

6. Draw **skeletal structures** for the following compounds:

(a) 2-chloro-3,3-dimethylheptanoic acid

(b) 3-bromo-3-ethyl-1-pentene