## CH1012 Tutorial 4

## Name:

- 1. Draw structures of the following transition metal compounds and if there are any isomers (geometric, optical, linkage) then sketch the isomers as well:
  - $Pd(NH_3)_2(CN)Cl$

• [Cd(H<sub>2</sub>O)(CO)BrF]

•  $[Rh(ox)_2(NH_3)_2]^{2-}$ 

2. Assign the oxidation state of carbon in the following compounds:  $C_{diamond}$  NaHCO<sub>3</sub>  $C_2F_2$ 

3. Determine if the following reactions are redox or metathesis reactions, explain your decision. For <u>redox</u> reaction(s) balance the equation and identify the oxidised and reduced species.

(i) 
$$2Ca(s) + O_2(g) \rightarrow 2CaO(s)$$

4. Balance the following redox equation in acidic solution, show all the steps in your working.

$$MnO_4^{\text{-}}(aq) \hspace{1cm} + \hspace{1cm} H_2O_2(aq) \hspace{3mm} \rightarrow \hspace{3mm} Mn^{2+}(aq) \hspace{3mm} + \hspace{3mm} O_2(g)$$

5. What is the **spectrochemical series** and how does this series help us understand the colour of transition metal complexes?