CH1012

Tutorial 1

Name:

1. Fuel desulfurization is often used in natural gas power plants to reduce one of the major pollutants resulting from fossil fuel combustion SO_2 . The compound removed from the fuel is hydrogen sulfide (H₂S).

 $2H_2S_{(g)} + O_{2(g)} \leftrightarrows 2S_{(s)} + 2H_2O_{(g)}$

Write down the equilibrium expression K_c for this reaction. What happens to the equilibrium concentration of H_2S if O_2 is added? What happens to the equilibrium concentration of S if H_2S is removed?

2. Write down the equilibrium constant (K_p) for the decomposition of liquid water for form gaseous hydrogen and oxygen.

3. What is the difference between a Lyophobic and a Lyophilic colloid – provide examples to illustrate your answer? Include a clear definition of each colloid.

4. The relative mobility of the hydrated Group 1A ions increases as we go down the group and yet the ionic size increases. Explain this apparent anomaly.

5. Define the term **organometallic compound** and give an example of a Group 2A organometallic compound.