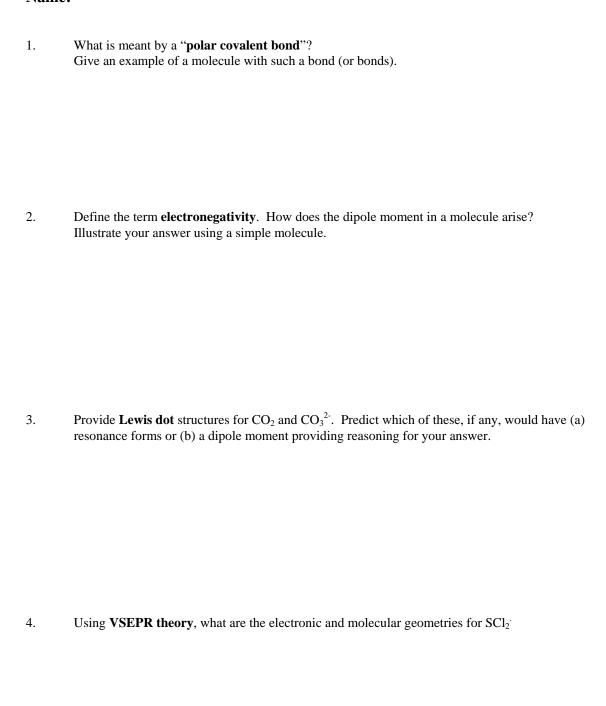
CH1011 Tutorial 2

Name:



5.	iron
6.	A lead acetate solution { $Pb(CH_3COO)_2$ } is reacted with sodium chloride solution to form a precipitate of lead(II) chloride. A solution of 1068 mL of 244g/L lead acetate is reacted with 500 mL of 0.850 M sodium chloride.
	► Write down the balanced equation.
	What is the concentration of the lead acetate solution in mol/L?How many kg of lead chloride are formed?