

1/9/06

①

$$\frac{C_1 V_1}{C_2 V_2} = \frac{x}{y}$$

4.80 How many mL of
0.120 M HCl are need
to neutralise 50.0 mL
of 0.101 M $\text{Ba}(\text{OH})_2$?
(base)

4.84 6.82 g of $\text{Sr}(\text{NO}_3)_2$ is
dissolved in 0.500 L of water.
0.100 L is drawn from this
solution and titrated with
0.0335 M Na_2CrO_4
what volume of Na_2CrO_4 is
needed to precipitate all the
 Sr^{2+} as SrCrO_4 ?